# Calcium-Catalyzed Asymmetric Synthesis of 3‑Tetrasubstituted Oxindoles: Efficient Construction of Adjacent Quaternary and Tertiary Chiral Centers

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# **S** Supporting Information

[AB](#page-3-0)STRACT: [Chiral Ca-cat](#page-3-0)alyzed asymmetric addition reactions of 3-substituted oxindoles with N-Boc-imines afford 3 tetrasubstituted oxindole derivatives bearing adjacent quaternary and tertiary chiral centers, which are key structures for biological activities. Ubiquitous and nontoxic Ca catalysts (1− 10 mol %) work well in this reaction, and high yields (up to 99%) and selectivities (up to >99% ee) of the products with wide substrate scope have been attained. The structures of the chiral Ca catalysts and intermediary Ca enolates are also discussed.

 $\sum_{\text{for several biological activities, such as anticancer,}^{1} \text{ anti-}$ <br>  $\sum_{\text{outivalential}}^{2}$  antimalarial  $3$  antitubarcular  $4$  proceeds are excepted  ${\rm HIV.}^2$  antimalarial, $^3$  antitubercular, $^4$  progesterone receptor agonist,<sup>5</sup> and MDM2 inhibitor,<sup>6</sup> etc.<sup>7</sup> For asymmetric sy[nt](#page-3-0)hesis of t[he](#page-3-0)se compoun[ds](#page-3-0), construction [o](#page-3-0)f the quaternary chiral centers [o](#page-3-0)ften observed in the[ir](#page-3-0) str[uc](#page-3-0)tures is one of the most challenging tasks.<sup>8</sup> While the reactions of 3-substituted oxindoles with imines potentially provide an efficient method for making adjacent quatern[ar](#page-3-0)y and tertiary chiral centers,<sup>9</sup> control of such stereogenic centers is known to be very difficult presumably because of steric congestion. Up to now, only fo[u](#page-3-0)r reports appear for this reaction, and yields, selectivities, catalyst efficiency, and/ or substrate generality are not satisfactory.<sup>10</sup> In 2008, Chen and co-workers reported the first example of the reaction of 3 substituted oxindoles with N-Boc-imin[es](#page-3-0) using bifunctional thiourea−tertiary amine organocatalysts, and high enantioselectivities (up to 95% ee) were obtained for several 3-substituted oxindoles with aromatic and heteroaromatic N-Boc-imines.<sup>10a</sup> However, a racemic compound was obtained in a reaction with an aliphatic N-Boc-imine. In 2009, Maruoka and co-work[ers](#page-3-0) reported phosphonium salts as chiral phase-transfer catalysts and demonstrated the reactions of 3-aryloxindoles with aromatic N-Boc-imines and good enantioselectivities (up to 88% ee) were obtained.10b Liu, Chen, and co-workers reported modified cinchona alkaloid-catalyzed reactions of N-unprotected 3 substitut[ed](#page-3-0) oxindoles with aromatic N-Ts-imines, and good enantioselectivities (up to 89% ee) were obtained.<sup>10c</sup> Finally, in 2013, Zhang and Peng reported the reactions of N-unprotected 3-bromooxindoles with aromatic N-Ts-imines [cat](#page-3-0)alyzed by bifunctional thiourea derivatives, and high enantioselectivities (up to 99% ee) were obtained.<sup>10d</sup> In those four reports, while high yields and high selectivities were obtained in some cases, substrates were limited to aro[matic](#page-3-0) and heteroaromatic imines,



and no enantioselective reactions with aliphatic imines were reported. In addition, organocatalysts (nonmetal catalysts) are used in all of the previous examples, and relatively high loading amounts of catalysts (10 mol %) are used except for phosphonium salts (3 mol %). Here, we report the reactions of 3-substituted oxindoles with N-Boc-imines using chiral calcium catalysts. Not only aromatic imines but also aliphatic imines work well, and high yields and high diastereo- and enantioselectivities have been attained using 1−10 mol % loadings of the calcium catalysts with wide substrate scope.

Recently, we have been interested in calcium compounds as catalysts in organic transformations. Calcium is a ubiquitous element, nontoxic, and readily available.<sup>11</sup> Previously, we have investigated chiral calcium alkoxide-catalyzed asymmetric reactions,<sup>12</sup> such as asymmetric Mic[ha](#page-3-0)el reactions,<sup>12a−e,h,i</sup> asymmetric  $[3+2]$  cycloaddition reactions,  $1^{2a, b,i}$  and asymmetric prot[on](#page-3-0)ation reactions.<sup>12f</sup> More recently, we have de[veloped](#page-3-0) chiral calcium chloride $^{12h,i}$  and chiral calciu[m iod](#page-3-0)ide  ${\rm (CaI_2)}^{13}$  for catalytic asymmetric [tra](#page-3-0)nsformations. We chose 3-methylsubstituted oxindole [1a](#page-3-0) [a](#page-3-0)nd N-Boc-imine 2a as models[, a](#page-3-0)nd several reaction conditions were examined. It was found that a combination of  $Cal_2$  with Pybox L gave promising results. When  $Cal<sub>2</sub>$  and L1 were used, the reaction proceeded smoothly to afford the desired product 3aa in 85% yield with high diastereoselectivity and moderate enantioselectivity (Table 1, entry 1). Ligand screening showed that L5 gave the best enantioselectivity (entries 2−5). With the optimized ligand L[5](#page-1-0), several solvents were screened. Slightly higher enantioselectivities were observed when diethyl ether  $(Et<sub>2</sub>O)$  and dichloromethane (DCM) were employed (entries 6−8). By lowering the

Received: March 17, 2015 Published: April 7, 2015

<span id="page-1-0"></span>Table 1. Optimization of Reaction Conditions



 $a$ Isolated yield.  $b$ Determined by  $1H$  NMR analysis of the crude reaction mixture. Contemnion by the training many of the crute  $\text{Area}$ added at  $-78$  °C.  ${}^{6}Et_{3}N$  was added at rt. *f*Slow addition of imine over 1 h; the reaction was further continued for  $17$  h.  $81a$  (1.0 equiv), 2a (1.5 equiv).  $h$ Slow addition of imine over 10 h; the reaction was continued for 8 h. <sup>*i*</sup>Slow addition of imine over 20 h; the reaction was continued for 22 h.  $j_{0.4}$  M.  $k_{CaCl<sub>2</sub>}$  was used instead of CaI<sub>2</sub>.  $l_{Ca(O-i)}$  $\text{Pr}_2$  was used instead of CaI<sub>2</sub> without Et<sub>3</sub>N.



reaction temperature, the enantioselectivity was further improved to 93% ee (entry 9). To decrease the catalyst loading, several conditions were investigated (entries 10−15). When the ratio of  $Cal<sub>2</sub>$  to L5 was 1:1.5, the desired product 3aa was obtained in 87% yield with 95:5 dr and 98% ee (entry 12). Further decreasing the catalyst loading showed promising results (entries 13−15). Finally, the reaction proceeded smoothly to afford the desired adduct in 97% yield with high diastereoselectivity  $(3aa/4aa = 94/6)$ , and the enantioselectivity of 3aa was 97% using 1 mol % of the catalyst (entry 15). On the other hand, while the enantioselectivity decreased slightly using  $CaCl<sub>2</sub>$ instead of  $Cal<sub>2</sub>$  (entry 16), a significant drop in the enantioselectivity was observed when calcium isopropoxide  $(Ca(O-i-Pr)_2)$  was used as a calcium source (entry 17).

Next, we investigated the substrate scope of this asymmetric reaction (Table 2). When we applied tolyl imines 2b−d, high yields, high diastereoselectivities, and excellent enantioselectivities were observed in all cases (entries 2−4). In the reactions

Table 2. Substrate Scope<sup>a</sup>

$\mathsf{R}^2$	1 R <sup>3</sup> 1.5 equiv	R <sup>1</sup> Boc Boc	$Cal2$ (5 mol %) $Et_3N$ (5 mol %) DCM (0.2 M) –78 °C, 18 h	$R^1$ Pybox L5 (7.5 mol %) $R^2$	$\mathsf{R}^3$ <sup>1</sup> NHBoc Boc $R^2$ 3	R <sup>1</sup> Boc 4	NHBoc
	entry	R <sup>1</sup>	R <sup>2</sup>	$R^3$	yield <sup>b</sup> (%)	$3:4^c$	$ee^d$ (% )
	$\mathbf{1}$	Me(1a)	Н	Ph $(2a)$	87 (3aa)	95:5	98
	$\overline{c}$	Me(1a)	Н	$o$ -Me $C_6H_4$ (2b)	93 (3ab)	97:3	97
	$3^e$	Me(1a)	Η	$m\text{-}MeC_6H_4(2c)$	80 (3ac)	95:5	97
	4 <sup>f</sup>	Me(1a)	Η	$p$ -MeC <sub>6</sub> H <sub>4</sub> (2d)	95 (3ad)	96:4	95
	5	Me(1a)	Н	$m$ -CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> (2e)	93 (3ae)	96:4	97
	6	Me(1a)	Η	$p$ -OMe C <sub>6</sub> H <sub>4</sub> (2f)	91(3af)	94:6	95
	7	Me(1a)	Η	$p\text{-}\mathrm{ClC}_6\mathrm{H}_4(2g)$	92 (3ag)	97:3	96
	8	Me $(a)$	Н	$2$ -furyl $(2h)$	89 (3ah)	97:3	96
	9	Me $(a)$	Η	$2$ -thienyl $(2i)$	91 (3ai)	98:2	98
	10	Me(1a)	Η	2-naphthyl $(2j)$	84 (3aj)	97:3	97
	$11^{\ensuremath{\textit{fs}}\xspace}$	Me $(a)$	Н	Et(2k)	96 (3ak)	97:3	90
	$12^{f,g,h,i}$	Me(1a)	Н	$n\text{-}C_5H_{11}$ (21)	95 (3al)	93:7	72
	$13^{e, h, i}$	Me $(a)$	Н	$i$ -Bu $(2m)$	99 (3am)	94:6	84
	14 <sup>e</sup>	Bn(1b)	Η	Ph(2a)	81 (3ba)	94:6	95
	$15^e$	Bn(1b)	Η	2-thienyl $(2i)$	84 (3bi)	96:4	98
	16	$n\text{-}Pr(1c)$	Н	Ph $(2a)$	83 (3ca)	94:6	98
	17	$n\text{-}Pr(1c)$	Н	Et(2k)	43 (3ck)	97:3	72
	$18^{e,h,i}$	$n\text{-}Pr(1c)$	Η	$i$ -Bu $(2m)$	93(3cm)	97:3	74
	19	$i$ -Pr $(1d)$	Η	Ph(2a)	80 (3da)	95:5	>99
	$20^{e,h,i}$	$i$ -Pr $(1d)$	Н	$i-Bu(2m)$	77(3dm)	95:5	73
	21	$1e^{j}$	Н	Ph $(2a)$	80 (3ea)	95:5	93
	$22^f$	$1e^j$	Η	$p$ -MeC <sub>6</sub> H <sub>4</sub> (2d)	82 (3ed)	94:6	95
	$23^f$	$1e^j$	Н	$p$ -ClC <sub>6</sub> H <sub>4</sub> (2g)	88 (3eg)	96:4	97
	$24^e$	$1e^j$	Η	$2$ -furyl $(2h)$	93 (3eh)	96:4	91
	25	1 <sup>k</sup>	Me	Ph $(2a)$	93 (3fa)	94:6	95
	$26^e$	$1\mathrm{g}^l$	F	Ph $(2a)$	82 (3ga)	94:6	93
$a_{\mathbf{A}}$		11		$\mathbf{1}$	$\cdot$ 1	$b_{T}$ 1.1	. 11

 $a_2$  was added over 1 h unless otherwise noted.  $b_1$  Isolated yield. Extermined by <sup>1</sup>H NMR analysis of the crude reaction mixture.<br><sup>d</sup>Determined by chiral HPLC analysis <sup>e</sup>Slow addition of imine over 10 Determined by chiral HPLC analysis. "Slow addition of imine over 10  $\frac{1}{2}$  becoming to  $\frac{1}{2}$  and  $\frac$  $\frac{1}{2}$  over 5 h; the reaction was further continued for 13 h.  ${}^{g}$ Imine (3 equiv) was used.  $h$  10 mol % of the catalyst was used.  $i$ Pybox L3 was used.  $j$ 1e:  $R_1 = CH_2CH(O(CH_2)_2O)$ ,  $R_2 = H_1$ .  $k_1$  = CH<sub>2</sub>CH(O(CH<sub>2</sub>)<sub>2</sub>O),  $R^2$  = Me.  ${}^{1}I$ **g**:  $R^1$  = CH<sub>2</sub>CH(O(CH<sub>2</sub>)<sub>2</sub>O),  $R^2$  = F.

with imines 2e, 2f, and 2g, bearing electron-withdrawing and electron-donating substituents at the benzene rings, the reactions also proceeded smoothly to afford the desired products 3ae, 3af, and 3ag in excellent yields with high diastereo- and enantioselectivities (entries 5−7). When heteroaromatic imines 2h and 2i were examined, excellent enantioselectivities were obtained (entries 8 and 9). The 2-naphthylaldehyde-derived imine 2j also gave excellent enantioselectivity (entry 10). For aliphatic imines, imine 2k derived from propionaldehyde reacted with oxindole 1a to afford the desired product 3ak in excellent yield with excellent diastereo- and enantioselectivities (entry 11).  $n-C<sub>5</sub>H<sub>11</sub>$  imine 2l and *i*-Bu imine 2m also reacted smoothly to provide the desired products 3al and 3am, respectively in excellent yields with good to high diastereo- and enantioselectivities using Pybox L3 as the ligand (entries 12 and 13). It should be noted that aliphatic imines reacted smoothly to give the desired adducts in excellent yields with good to excellent diastereo- and enantioselectivies. In previous reports, as mentioned, there were no examples of enantioselective reactions using aliphatic imines.<sup>10</sup> The substrate scope of oxindoles 1 was also surveyed. 3-Benzyl-substituted oxindole 1b reacted with imines 2a and 2i smo[oth](#page-3-0)ly to afford the desired adducts in high yields with excellent diastereo- and enantioselectivities (entries 14 and 15). The absolute configuration was determined by comparing the optical rotation of 3bi with that of the literature.<sup>10a</sup> 3-n-Propyl- and 3-isopropyl-substituted oxindoles 1c and 1d also reacted with imine 2a smoothly, and almost perfect e[nan](#page-3-0)tioselectivities were obtained (entries 16 and 19). Aliphatic imines 2k and 2m also worked well (entries 17, 18, and 20). We also examined the reactions of substituted oxindoles 1e, 1f, and 1g (entries 21−26).<sup>14</sup> In all cases, the reactions proceeded smoothly to afford the desired adducts in high yields with excellent diastereo- and en[an](#page-3-0)tioselectivities. The adduct 3ea was readily converted to spirooxindole derivative 5ea in good yield without loss of any optical purity (Scheme 1).<sup>15−17</sup> Thus, we

#### Scheme 1. Transformation to Spirooxindole



have succeeded in providing oxindole derivatives that have adjacent quaternary and tertiary chiral centers in satisfactory yields and stereoselectivities with wide substrate scope.

An assumed catalytic cycle of this asymmetric reaction is shown in Scheme 2. CaI<sub>2</sub> and ligand L5 may form a 1:1 complex (6) and a 1:2 complex in solution; that is supported by NMR experiments.<sup>13,18</sup> Chiral Ca catalyst 6 may work as a monomeric form because a linear correlation between ligand ee's and product ee's was obs[erved](#page-3-0).<sup>18,19</sup> Ca catalyst 6 activates oxindole 1 to form Ca enolate 7, which was detected by ESI HRMS analysis.<sup>18</sup> One enantioface (the [Re](#page-3-0) [fac](#page-3-0)e) of 7 is effectively shielded by the chiral



a Iodide ions are omitted for clarification.

ligand (vide infra), and 7 attacks imine 2's Re face smoothly to give intermediate 8. There may be two possible ways to complete the catalytic cycle. In the first, the intermediate 8 is protonated by  $\mathrm{Et}_3\mathrm{NH}^+ \mathrm{I}^-$  to give product 3 along with regeneration of the chiral Ca catalyst 6 (path A). In the second, the intermediate 8 deprotonates 1 to form the Ca enolate 7 (path B). Given the steric bulkiness of 1 and 8, path A may be more likely; however, path B cannot be ruled out at this moment. It is noteworthy that the Ca catalysts, which have both mild Lewis acidity and Brønsted basicity,  $11$  may play a key role in obtaining high yields and stereoselectivities and also efficient catalyst turnover.

We further inv[est](#page-3-0)igated stereochemical structures of the chiral Ca catalyst and the Ca enolate. While X-ray crystallographical analysis of  $Ca(CIO<sub>4</sub>)<sub>2</sub>$ −Pybox was conducted,<sup>20</sup> we carried out DFT calculation of CaCl<sub>2</sub>−L5 (Scheme 3).<sup>21</sup> The optimized

#### Scheme 3. Stereochemical Model of CaCl<sub>2</sub>−[L](#page-3-0)5



Blue N, Red O, Gray C, Green Cl, Yellow Ca; a:  $3.8 \text{ Å}$ ; b:  $2.6 \text{ Å}$ ; c:  $2.7 \text{ Å}$ 

structure has  $C_2$ -symmetry, and the three nitrogen atoms of L5 coordinate to Ca. On the other hand, the length between Ca and O of the methoxy group of L5 is 3.8  $\AA$ , suggesting weak coordination of O to Ca. We also investigated the structure of the Ca enolate (Scheme 4). The DFT calculation showed that the

Scheme 4. Stereochemical Model of Ca Enolate



three nitrogen atoms and the carbonyl oxygen of the Boc group coordinated to Ca. The length between Ca and O of the methoxy groups of L5 is 3.9 and 4.0 Å, also suggesting weak interactions between O and Ca. The stereochemical model of the Ca enolate indicates that the plane of chiral ligand L5 and the plane of the enolate are almost perpendicular and that the Re face of the enolate is efficiently shielded by the  $CH<sub>2</sub>OMe$  arm of L5.

In conclusion, we have developed a new catalytic method for the synthesis of optically active 3-tetrasubstituted oxindoles using chiral Ca catalysts. Adjacent quaternary and tertiary chiral centers have been created efficiently by the Ca-catalyzed reactions of 3 substituted oxindoles with imines. High diastereo- and enantioselectivities have been realized with wide substrate scope. This is the first example of a chiral metal-catalyzed reaction of a 3-substituted oxindole with an imine. The

<span id="page-3-0"></span>ubiquitous and nontoxic Ca catalyst worked well, and even 1 mol % of the catalyst gave a high yield and high selectivities of the product. This reaction may provide an efficient approach to the synthesis of biologically important oxindoles, including spirooxindole derivatives. Further studies on this topic are underway in our laboratories.

# ■ ASSOCIATED CONTENT

#### **S** Supporting Information

Experimental section and copies of NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

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#### **Notes**

The authors declare no competing financial interest.

#### ■ ACKNOWLEDGMENTS

This work was partially supported by a Grant-in-Aid for Science Research from the Japan Society for the Promotion of Science (JSPS), Global COE Program, The University of Tokyo, MEXT, Japan, Japan Science Technology Agency (JST). We thank Mr. K. Takasugi (The University of Tokyo and JEOL Ltd.) for helpful discussions about NMR experiments.

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(18) See the Supporting Information.

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(20) Yamamura, M.; Miyake, J.; Imamura, Y.; Nabeshima, T. Chem. Commun. 2011, 47, 6801.

 $(21)$  Because of the difficulty of calculation, we used CaCl<sub>2</sub> instead of CaI<sub>2</sub>. CaCl<sub>2</sub> also showed a high yield and high selectivities (see Table 1, entry 16).